## 1101

## F.E. (Sem-II) (All Branches) (Rev.) (CBSGS)

Paper / Subject Code: 29603 / Applied Chemistry- II.

[Time: 2 Hours]

Please check whether you have got the right question paper.

Date-29/11/18

[ Marks: 60]

Q.P. Code:38692

|    | N.B:           | <ol> <li>Question No.1 is Compulsory.</li> <li>Attempt any three questions from remaining five questions.</li> <li>Figures to the right indicate Full marks.</li> <li>All questions carry equal marks.</li> <li>Atomic weights: - H=1, C=12, N=14, O=16, S=32, CI=35.5, Ba=137.3, Ca=40. Mg=24, Na=23.</li> </ol> |    |
|----|----------------|---|----|
| 1. | Answe          | r any five from the following:-   | 15 |
|    | b)<br>c)<br>d) | Distinguish between anodic and cathodic coatings for corrosion prevention.  What are Fuels? Give its classification.  Give Composition, Properties and Uses of <b>magnalumin</b> .  What are composite materials? Mention its applications.  Explain the green chemistry principal 'Accident prevention'.         |    |
|    | f)             | Explain with example, how the nature of volatile oxide film formed on the surface of metal influences the rate of corrosion.  1.5 g of the same coal sample in a Bomb-calorimeter experiment gave 0.36 g BaSO <sub>4</sub> . Calculate percentage of S in the coal sample.  |    |
| 2. |                | Explain the mechanism of following types of corrosion:  i) Galvanic corrosion  ii) Waterline corrosion  Define cracking of hydrocarbons. Explain fixed bed catalytic cracking.  | 06 |
|    |                | Calculate % Atom Economy for the following reaction   | 05 |
|    |                | $C_6H_6 + Cl_2 \longrightarrow C_6H_5CI + HCI$  | 04 |
| 3. | a)             | A fuel sample has the following composition: $C_2H_4 = 38\%$ , $C_2H_6 = 14\%$ , $CO = 7\%$ , water vapour = 2.0% and rest is nitrogen. Calculate the volume of oxygen and air required for complete combustion of $5m^3$ of fuel.  | 06 |
|    | b)             | Explain Conventional and Greener route for synthesis of Adipic acid. Mention the green Chemistry principle involved.  | 05 |
|    | c)             | How do the following factors related to nature of environment affect corrosion?  i) Relative area of anode and cathode  ii) Temperature   | 04 |
| 4. | a)             | What are nonferrous alloys? Distinguish between brass and bronze.   | 06 |
|    |                | What is the principle of cathodic protection? Explain impressed current protection method.  | 05 |
|    | c)             | Write a note on sandwich panel composites.  | 04 |

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04

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- 5. a) What is Biodiesel? Give 'Trans-esterification', reaction to obtain Biodiesel from **06** vegetable oils. Mention its advantages.
  - b) What is powder metallurgy? Explain Mechanical Pulverisation and Atomization **05** methods for manufacturing metal powders.
  - c) What are the important properties of composite materials?
- a) Mention methods for applying metal coatings on the metals? Give brief account of **05** Metal cladding
  - b) A coal sample contains, C=82%, O=6%, H=4%, S=1.5%, N=1% and Ash=5.5%. **05** Calculate the GCV and NCV of given coal sample.
  - c) Discuss the various steps involved in powder metallurgy and mention its application. 05